## New Secoiridoid Glucosides from Swertia japonica

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Four new secoiridoid glucosides, swertiajaposides  $C - F (1 - 4, \text{resp.})$ , were isolated from the whole plant of Swertia japonica Makino together with two known compounds, 8-hydroxy-10-hydrosweroside (5) and senburiside IV (6). The structures of  $1 - 4$  were elucidated on the basis of spectroscopic, chemical, and physicochemical evidence.

Introduction. – The whole plants of Swertia japonica Makino (Gentianaceae) are the crude drug Swertia Herb, used as a stomachic or stimulant of appetite in Japan [1]. The constituents of this crude drug have previously been investigated and shown to contain secoiridoid glucosides  $[2-6]$ , xanthones  $[7-10]$ , flavonoids  $[11]$ , biphenyl glucosides [12], triterpenoids [9] [13], and 2,8-dioxabicyclo[3.3.1]nonanes [14]. In previous works we reported the structure determination of ten new secoiridoid glucosides  $[15][16]$ , a new unsaturated alcohol glucoside  $[16]$ , and a new lignan glucoside [16] from the whole plant of S. japonica. Here, we report the isolation and structure elucidation of four new secoiridoid glucosides, swertiajaposides  $C - F(1-4)$ , resp.), together with two known compounds, 8-hydroxy-10-hydrosweroside (5) and senburiside IV  $(6)$ , from the whole plant of S. *japonica*.

Results and Discussion. – The dried whole plants of S. japonica were extracted with MeOH. The MeOH extract was partitioned between  $H_2O$  and CHCl<sub>3</sub>,  $H_2O$  and Et<sub>2</sub>O, H2O and AcOEt, and H2O and BuOH. The BuOH-soluble fraction was subjected to column chromatography (silica gel and *Sephadex LH-20*) and preparative HPLC to afford two new compounds, named swertiajaposide C  $(1)$  and swertiajaposide F  $(4)$ , and two known compounds, 8-hydroxy-10-hydrosweroside (5) and senburiside IV (6). The H<sub>2</sub>O-soluble fraction was subjected to column chromatography (*Diaion HP-20* and silica gel) and preparative HPLC to afford two new compounds, named swertiajaposide  $D(2)$  and swertiajaposide  $E(3)$ .

Swertiajaposide C (1) was obtained as an amorphous powder. The molecular formula of 1 was determined as  $C_{17}H_{24}O_{10}$  based on the positive HR-FAB-MS ( $m/z$ 389.1442 ( $[M + H]$ <sup>+</sup>, calc. 389.1447)). Acid hydrolysis of 1 gave p-glucose, which was identified by its retention time and optical rotation using chiral detection by HPLC analysis. The <sup>1</sup>H-NMR spectrum of  $1$  (in CD<sub>3</sub>OD; *Table 1*) exhibited signals due to one Me group ( $\delta(H)$  1.97 (d, J = 7.3)), two CH<sub>2</sub> (one oxygenated), one MeO group ( $\delta(H)$ ) 3.58 (s)), one oxygenated CH ( $\delta$ (H) 5.38 (d, J = 1.2)) and one C = C–H moiety.

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Furthermore, one anomeric signal ( $\delta$ (H) 4.83 (d, J = 7.8)) was recognized. The coupling constant of the anomeric signal indicated that the glucosyl linkage has  $\beta$ -configuration. The <sup>13</sup>C-NMR spectrum of 1 (in CD<sub>3</sub>OD; *Table 1*) showed signals due to one fully substituted C=C bond ( $\delta$ (C) 119.7, 147.3) and one C=O group ( $\delta$ (C) 165.7). The <sup>1</sup>H,<sup>1</sup>H-COSY spectrum of **1** (*Fig. 1*) implied connectivities of CH<sub>2</sub>(6) to CH<sub>2</sub>(7)<sup>1</sup>), and of  $H-C(8)$  to Me(10). The HMBC spectrum (*Fig. 1*) showed correlations between  $H-C(1)$  and  $C(5)$  and  $C(8)$ , between  $H-C(3)$  and  $C(1)$ ,  $C(5)$ , and  $C(11)$ , between  $CH<sub>2</sub>(6)$  and C(4), between CH<sub>2</sub>(7) and C(11), between H-C(8) and C(5), between  $H-C(1')$  and  $C(1)$ , and between MeO and  $C(3)$ , respectively. With these correlations, the constitution of 1 could be deduced. The relative configuration of 1 was determined by NOESY experiments. The NOE cross-peaks observed between  $H_a-C(1)$  and  $H_a-C(3)$ , and between the MeO group at C(3) and H-C(1') implied that the Glc moiety at C(1) and the MeO group at C(3) occurred on the same face  $(\beta)$  of the ring system (Fig. 2). The  $C(8) = C(9)$  bond was (Z)-configured, based on a NOESY crosspeak between  $H<sub>a</sub> - C(1)$  and Me(10) (Fig. 2). From these data, the structure of 1 was

<sup>1)</sup> Arbitrary atom numbering, see Formula collection.

Position	$\delta(H)$	$\delta(C)$
1	6.22 $(d, J=0.7)$	91.2
3	5.38 $(d, J = 1.2)$	95.9
4		119.7
5		147.3
6	2.66 (ddd, J = 17.3, 4.4, 4.0, H <sub><math>_{\beta}</math></sub> ), 2.76 (dddd, J = 17.3, 11.0, 5.6, 1.0, H <sub>a</sub> )	23.8
	4.36 (ddd, $J = 11.2, 11.0, 4.4, H_8$ ), 4.48 (ddd, $J = 11.2, 5.6, 4.0, H_a$ )	66.8
8	6.42 $(q, J = 7.3)$	136.9
9		130.6
10	1.97 $(d, J = 7.3)$	14.4
11		165.7
1'	4.83 $(d, J = 7.8)$	99.2
$2^{\prime}$	3.20 (dd, $J = 9.0, 7.8$ )	75.0
3'	3.45 $(t, J = 9.0)$	78.6
4'	<sup>a</sup>	71.8
5'	$3.37 - 3.41$ ( <i>m</i> )	78.0
$6^{\prime}$	3.68 (dd, $J = 11.7, 6.1$ ), 3.92 (dd, $J = 11.7, 2.2$ )	63.0
$3-MeO$	3.58(s)	58.2

Table 1. <sup>*IH*</sup>- and <sup>13</sup>C-NMR Spectral Data of 1. At 400/100 MHz, resp., in CD<sub>3</sub>OD;  $\delta$  in ppm, *J* in Hz.



elucidated as  $(5Z, 6S^*, 8S^*)$ -5-ethylidene-6- $(\beta$ -D-glucopyranosyloxy)-4,5,6,8-tetrahydro-8-methoxy-1H,3H-pyrano[3,4-c]pyran-1-one, the absolute configuration of which remains to be established.

Swertiajaposides D (2) and E (3) gave the same molecular formula,  $C_{16}H_{24}O_9$ , by positive HR-FAB-MS (2,  $m/z$  383.1293 ( $[M + Na]$ <sup>+</sup>, calc. 383.1318); 3,  $m/z$  383.1288  $([M+Na]^+,$  calc. 383.1318)). Acid hydrolysis of 2 and 3 each gave D-glucose, as described above for 1. Both compounds showed closely similar <sup>1</sup>H- and <sup>13</sup>C-NMR spectral features (in CD<sub>3</sub>OD; Table 2). In the <sup>1</sup>H-NMR spectrum, each compound exhibited signals due to one Me group, two CH (one oxygenated) and four  $\text{CH}_2$  (three oxygenated). Furthermore, one anomeric signal  $(2, \delta(H) 4.22 (d, J = 7.8); 3, \delta(H) 4.26$ 



Fig. 2. NOESY (full-line arrows) Correlations for 1

Position	2		3	
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$
1	3.78 (dd, $J = 10.5$ , 4.4, H <sub>a</sub> ),	67.6	3.67 (dd, $J = 10.0, 5.6, Ha$ ),	68.3
	4.08 (dd, $J = 10.5$ , 3.4, $H_h$ )		4.26 (dd, $J = 10.0, 5.9, H_h$ )	
3	4.21 (br. $d, J = 15.9, H_\beta$ ),	63.7	4.28 (br. $d, J = 15.4, H_a$ ),	65.6
	4.30 (br. d, $J=15.9$ , H <sub>a</sub> )		4.38 (br. d, $J=15.4$ , H <sub>b</sub> )	
$\overline{4}$		125.1		124.4
5		153.5		155.1
6	$2.37 - 2.41$ ( <i>m</i> , H <sub><i>a</i></sub> ),	26.9	2.44 – 2.46 $(m, H_6)$ ,	28.8
	$2.78 - 2.86$ ( <i>m</i> , H <sub>a</sub> )		2.79 – 2.85 $(m, H_a)$	
7	4.37 (ddd, $J = 11.0, 10.5, 4.4, H_8$ ),	67.8	$4.39 - 4.42$ ( <i>m</i> , 2 H)	67.9
	4.43 (ddd, $J = 11.0, 5.4, 5.4, H_a$ )			
8	3.83 $(qd, J=6.3, 6.3)$	71.8	3.79 $\left( qd, J=6.6, 2.9 \right)$	72.7
9	$2.35 - 2.37$ ( <i>m</i> )	46.5	$2.44 - 2.46$ ( <i>m</i> )	43.8
10	1.31 $(d, J=6.3)$	19.3	1.30 $(d, J=6.6)$	18.1
11		166.0		165.7
1'	4.22 $(d, J=7.8)$	104.6	4.26 $(d, J = 7.8)$	102.0
$2^{\prime}$	3.13 (dd, $J = 9.0, 7.8$ )	74.9	3.15 $(dd, J=9.3, 7.8)$	75.1
3'	$^{a}$ )	78.2	$^{a}$ )	78.1
4'	$^{a})$	71.7	$^{a})$	71.7
5'	a)	78.1	a)	78.1
$6^{\prime}$	3.66 (dd, $J = 11.7, 5.6$ ),	62.8	3.65 (dd, $J = 12.0, 5.6$ ),	62.9
	3.87 $(dd, J=11.7, 1.5)$		3.88 $(dd, J=12.0, 1.7)$	

Table 2. <sup>*IH*</sup>- and <sup>13</sup>C-NMR Spectral Data of 2 and 3. At 400/100 MHz, resp., in CD<sub>3</sub>OD;  $\delta$  in ppm, J in Hz.

a) Overlapped by the solvent signal.

 $(d, J = 7.8)$ ) was recognized. The coupling constant of the anomeric signal indicated that the glucosyl linkage has  $\beta$ -configuration. In the <sup>13</sup>C-NMR spectrum, each compound showed signals due to one fully substituted C=C bond (2,  $\delta$ (C) 125.1, 153.5; 3,  $\delta$ (C) 124.4, 155.1) and one C=O group (2,  $\delta$ (C) 166.0; 3,  $\delta$ (C) 165.7). The <sup>1</sup>H,<sup>1</sup>H-COSY spectrum of 2 and 3 (Fig. 3) implied connectivities of CH<sub>2</sub>(1) to H – C(9), of CH<sub>2</sub>(6) to CH<sub>2</sub>(7), of H – C(8) to H – C(9), and of H – C(8) to Me(10). The HMBC spectrum of 2 and 3 (Fig. 3) showed correlations between  $H-C(1)$  and  $C(5)$ , between  $H-C(3)$  and C(5), between CH<sub>2</sub>(6) and C(4), between CH<sub>2</sub>(7) and C(11), between H–C(8) and  $C(3)$  and  $C(5)$ , and between  $H-C(1')$  and  $C(1)$ . From these data, the constitutional



Fig. 3. <sup>1</sup>H,<sup>1</sup>H-COSY (bold line) and HMBC (full-line arrows) correlations for 2 and 3

formulae of 2 and 3 could be deduced. The relative configuration of 2 and 3 was determined as follows. In the <sup>1</sup>H-NMR spectrum of  $2, H-C(8)$  has a large coupling constant ( $J(8\beta, 9\alpha) = 6.3$ ). In an NOE experiment, irradiation of H – C(8) enhanced the signals of H<sub>b</sub>-C(1) (1.46%) and H<sub>b</sub>-C(3) (3.94%), and irradiation of Me(10) enhanced the signal of  $H_b-C(1)$  (1.01%). These observations revealed that 2 takes a conformation as shown in Fig. 4, and Me(10) and the  $(\beta$ -D-glucopyranosyloxy)methyl group at  $C(9)$  were on the  $\alpha$ - and  $\beta$ -faces of the ring system, respectively (*Fig. 4*). On the other hand, in the <sup>1</sup>H-NMR spectrum of 3,  $H - C(8)$  has a small coupling constant  $(J(8\alpha, 9\alpha) = 2.9)$ . In an NOE experiment, irradiation of H – C(8) enhanced the signal of  $H_a-C(3)$  (7.83%), and irradiation of Me(10) enhanced the signals of  $H_a-C(1)$  (1.32%) and  $H_b-C(1)$  (2.31%). These observations revealed that 3 takes a conformation as shown in Fig. 5, and Me(10) and the ( $\beta$ -D-glucopyranosyloxy)methyl group at C(9) occurred on the same face  $(\beta)$  of the ring system (Fig. 5). Accordingly, the structures of 2 and 3 were assigned as  $(5R^*, 6R^*)$ -5- $[(\beta$ -D-glucopyranosyloxy)methyl]-4,5,6,8-tetrahydro-6-methyl-1H,3H-pyrano[3,4-c]pyran-1-one and  $(5R*,6S^*)$ -5-[ $(\beta$ -D-glucopyrano-



Fig. 4. Selected coupling constant (dotted arrows) and NOEs (full-line arrows) for 2



Fig. 5. Selected coupling constant (dotted arrows) and NOEs (full-line arrows) for 3

syloxy)methyl]-4,5,6,8-tetrahydro-6-methyl-1H,3H-pyrano[3,4-c]pyran-1-one, respectively. The absolute configuration of 2 and 3 could not be determined yet.

Swertiajaposide F (4) was obtained as an amorphous powder. Acid hydrolysis of 4 gave d-glucose as described above. Compound 4 showed a very similar signal pattern to that of 2 in the <sup>13</sup>C-NMR spectrum (in CD<sub>3</sub>OD; *Table 3*). However, in contrast to 2, one more oxygenated CH signal was observed instead of a  $CH<sub>2</sub>$  signal. The molecular formula was determined as  $C_{16}H_{24}O_{10}$  based on the positive HR-FAB-MS ( $m/z$  377.1460  $([M + H]^+,$  calc. 377.1448). Consequently, 4 was deduced to be a compound in which a H-atom in 2 was replaced by an OH group. The resonance for C(6) at  $\delta$ (C) 26.9 of 2 was shifted downfield to  $\delta(C)$  62.1 in 4, suggesting that an additional OH group was located at  $C(6)$ . This was confirmed by the  ${}^{1}H, {}^{1}H$ -COSY spectrum in which a crosspeak was observed between  $H-C(6)$  and  $CH<sub>2</sub>(7)$ . The relative configuration of the OH group at  $C(6)$  was determined to be  $\beta$  from an NOE experiment, in which irradiation of  $H_a-C(9)$  enhanced the signal of  $H_a-C(6)$  (4.97%). The absolute configuration of 4 was determined as  $(6R, 8R, 9R)^1$ , based on the circular dichroism (CD) spectrum, in which a negative *Cotton* effect was observed at 231 nm  $(\Delta \epsilon = -7.20)$  [17] [18]. Accordingly, the structure of 4 was elucidated as  $(4R, 5R, 6R)$ -5- $[(\beta$ -D-glucopyranosyloxy)methyl]-4,5,6,8-tetrahydro-4-hydroxy-6-methyl-1H,3H-pyrano[3,4-c]pyran-1-one.

Position	$\delta(H)$	$\delta(C)$
$\mathbf{1}$	3.87 $(dd, J=10.2, 5.4)$ ,	69.2
	4.07 (dd, $J=10.2, 3.9$ )	
3	4.17 (dt, $J = 16.6$ , 2.0, H <sub>a</sub> ),	62.0
	4.38 (dd, $J = 16.6$ , 2.0, H <sub>a</sub> )	
4		126.3
5		152.3
6	$4.46 - 4.49(m)$	62.1
	4.32 (dd, $J = 11.5$ , 3.7, H <sub>b</sub> ), 4.50 (dd, $J = 11.5$ , 3.7, H <sub>a</sub> )	72.9
8	3.97 $\left( qd, J=6.3, 6.3 \right)$	71.0
9	$2.55 - 2.59(m)$	42.9
10	1.30 $(d, J=6.3)$	18.6
11		164.5
1'	4.26 $(d, J = 7.8)$	104.6
$2^{\prime}$	3.15 $(dd, J=9.0, 7.8)$	74.6
3'	$^{a}$ )	78.2
4'	$^{a}$ )	71.6
5'	a)	78.1
$6^{\prime}$	3.66 (dd, $J = 11.7, 5.4$ ), 3.86 (dd, $J = 11.7, 3.2$ )	62.8

Table 3. <sup>1</sup>H- and <sup>13</sup>C-NMR Spectral Data of 4. At 400/100 MHz, resp., in CD<sub>3</sub>OD;  $\delta$  in ppm, *J* in Hz.

The known compounds 5 and 6 were identified on the basis of their optical rotation values, NMR, and MS data as 8-hydroxy-10-hydrosweroside [19] [20] and senburiside IV [21] [22], respectively. This is the first report of compounds 5 and 6 from S. japonica.

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## Experimental Part

General. Column chromatography (CC): silica gel (SiO<sub>2</sub>) (230–400 mesh; Merck), Diaion HP-20 (250 – 300 mm; Mitsubishi Chemical Corporation), and Sephadex LH-20 (18 – 111 mm; GE Healthcare Bio-SciencesAB). HPLC: CCPM pump (Tosoh), RI-8020 refractive-index detector (Tosoh), UV-8020 UV/VIS detector (Tosoh) and JASCO OR-2090 plus chiral detector. Optical rotations: JASCO DIP-360 digital polarimeter. CD spectra: *JASCO J-720* spectropolarimeter;  $\Delta \varepsilon$  in mdeg l mol<sup>-1</sup> cm<sup>-1</sup> ( $\lambda$  in nm). UV Spectra: Beckman DU-64 spectrophotometer. IR Spectra: Perkin-Elmer Spectrum-One-FT-IR spectrometer. NMR Spectra: *JEOL JNM-LA 400* (<sup>1</sup>H, 400 MHz; <sup>13</sup>C, 100 MHz) spectrometer; chemical shifts  $\delta$  in ppm, rel. to Me<sub>4</sub>Si, J in Hz. FAB- and HR-FAB-MS: JEOL JMS-DX 303 mass spectrometer; glycerol as matrix; in  $m/z$ .

Plant Material. The dried whole plants of Swertia japonica were purchased from Uchida Wakanyaku Co. Ltd., Japan, in 2002. A voucher specimen (SJ-2002-01) was deposited at the Laboratory of Molecular Structural Analysis, Tohoku Pharmaceutical University.

Extraction and Isolation. The dried whole plants (including roots, stems, leaves and flowers) of Swertia japonica (2.0 kg) were extracted three times (14 d each time) with MeOH (6 l) at r.t. and filtered. The MeOH extract was concentrated under reduced pressure, and the residue  $(474 g)$  was suspended in H<sub>2</sub>O (11). This suspension was extracted with CHCl<sub>3</sub> ( $3 \times 11$ ), Et<sub>2</sub>O ( $3 \times 11$ ), AcOEt ( $3 \times 11$ ), and BuOH  $(3 \times 11)$ .

The BuOH-soluble fraction was concentrated under reduced pressure to afford a residue  $(12.0 g)$ , which was subjected to CC (SiO<sub>2</sub>, CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 30:10:1): 92 fractions according to TLC. Fr. 25, on prep. HPLC (TSKgel ODS-120T column (300 × 7.8 mm, 10 µm, Tosoh); MeOH/H<sub>2</sub>O 1:3, 1.0 ml/min) gave 1 (2.6 mg,  $t_R$  37.3 min). Fr. 42 on prep. HPLC (*Cosmosil 5C18AR* column (250  $\times$  10 mm, 10  $\mu$ m, *Nacalai Tesque*); MeOH/H<sub>2</sub>O 1:8, 1.0 ml/min) gave 4 (10.5 mg,  $t_R$  32.0 min) and 5 (3.5 mg,  $t_R$  38.4 min). Fr. 55 was purified over Sephadex LH-20 (MeOH/H<sub>2</sub>O 1:1) to afford 6 (113.7 mg).

The H<sub>2</sub>O-soluble fraction was passed through *Diaion HP-20* column, and the adsorbed material was eluted with H<sub>2</sub>O and MeOH. The MeOH eluate fraction was concentrated under reduced pressure to afford a residue (20.8 g), which was subjected to CC (SiO<sub>2</sub>, CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 30:10:1): 12 fractions according to TLC. Fr. 3 was further purified on prep. HPLC (*Cosmosil* 5SL column (250  $\times$  10 mm, 10 µm, *Nacalai Tesque*); CH<sub>2</sub>Cl<sub>2</sub>/MeOH/H<sub>2</sub>O 30:10:1, 1.0 ml/min) and gave 2 (10.4 mg,  $t_R$  24.6 min) and 3  $(26.1 \text{ mg}, t_R 29.2 \text{ min}).$ 

Swertiajaposide C  $(=(5Z,6S*,8S*)-5-Ethylidene-6-(\beta-D-glucopy ranosyloxy)-4,5,6,8-tetrahydro-8-1)$ methoxy-IH,3H-pyrano[3,4-c]pyran-1-one; 1). Amorphous powder.  $[\alpha]_{D}^{25} = -50.6$  (c=0.27, MeOH). UV (MeOH): 270 (4.3). IR (KBr): 3450, 1710. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 1*. FAB-MS (pos.): 389  $([M+H]^+)$ . HR-FAB-MS (pos.): 389.1442 ( $[M+H]^+$ , C<sub>17</sub>H<sub>25</sub>O<sub>10</sub>; calc. 389.1447).

Swertiajaposide  $D = (5R*, 6R*)-5-\int (\beta-D-Glucopyranosyloxy)methyl-4,5,6,8-tetrahydro-6-methyl-4,7,6,8-terahydro-6-methyl-4,7,8,8-terahydro-6-4,7,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-4,7,8,8-terahyl-$ *I*H,3H-pyrano[3,4-c]pyran-1-one; 2). Amorphous powder.  $\left[ \alpha \right]_{0}^{27} = +34.3$  (c = 1.04, MeOH). UV (MeOH): 224 (3.9). IR (KBr): 3445, 1725. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 2*. FAB-MS (pos.): 383  $([M + Na]^+)$ . HR-FAB-MS (pos.): 383.1293 ( $[M + Na]^+$ , C<sub>16</sub>H<sub>24</sub>NaO<sub>9</sub><sup>+</sup>; calc. 383.1318).

Swertiajaposide E  $(=(5R*,6S*)-5-\{(\beta-D-Glucopy ranosyloxy)methyl]-4,5,6,8-tetrahydro-6-methyl-$ *I*H,3H-pyrano[3,4-c]pyran-1-one; 3). Amorphous powder.  $[\alpha]_D^{27} = +74.1$  (c = 2.61, MeOH). UV (MeOH): 223 (3.9). IR (KBr): 3440, 1725. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 2*. FAB-MS (pos.): 383  $([M + Na]^+)$ . HR-FAB-MS (pos.): 383.1288 ( $[M + Na]^+$ , C<sub>16</sub>H<sub>24</sub>NaO<sub>9</sub><sup>+</sup>; calc. 383.1318).

Swertiajaposide F  $(=(4R,5R,6R)-5-[(\beta-D-Glucopyranosyloxy)methvl-4,5,6,8-tetrahvdro-4-hy-6]$ droxy-6-methyl-1H,3H-pyrano[3,4-c]pyran-1-one; 4). Amorphous powder.  $\left[ \alpha \right]_D^{25} = -26.3$  (c = 1.05, MeOH). CD (MeOH): +0.75 (262), -7.20 (231). UV (MeOH): 217 (4.0), 279 (2.8). IR (KBr): 3455, 1720. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 3*. FAB-MS (pos.): 377 ( $[M+H]^+$ ). HR-FAB-MS (pos.): 377.1460 ([ $M + H$ ]<sup>+</sup>, C<sub>16</sub>H<sub>25</sub>O<sub>10</sub>; calc. 377.1448).

8-Hydroxy-10-hydrosweroside  $(=(4aS,5R,6S)$ -5-Ethenyl-6-( $\beta$ -D-glucopyranosyloxy)-4,4a,5,6-tetra $hydro-IH, 3H-pyrano[3,4-cJpyran-1-one; 5).$   $[\alpha]_D^{24} = -150.7$  (c=0.35, MeOH). UV (MeOH): 243 (3.8). <sup>1</sup>H- and <sup>13</sup>C-NMR: as reported [19]. FAB-MS (pos.): 377 ( $[M+H]^+$ ).

Senburiside IV  $(=3'''-O-Glucosy/senburiside II; (IS,4aS,6R,7R,7aS)-1,4a,5,6,7,7a-Hexahydro-I-$ (hexopyranosyloxy)-6-[ (3-{[3-(hexopyranosyloxy)benzoyl]oxy}benzoyl)oxy]-7-methylcyclopenta[c]-

*pyran-4-carboxylic Acid*; 6).  $\left[a\right]_D^{25} = -80.5$  (*c* = 1.10, MeOH). <sup>1</sup>H- and <sup>13</sup>C-NMR: as reported [21] [22]. FAB-MS (pos.): 779 ( $[M + H]$ <sup>+</sup>).

Acid Hydrolysis of  $1-4$  and Determination of the Absolute Configuration of the Sugar. Each of the compounds,  $1-4$  (ca. 0.3 mg), was refluxed with 5% HCl for 5 h. The reaction mixture was neutralized with  $Ag_2CO_3$  and filtered. The soln. was concentrated in vacuo and dried to give a sugar fraction. The sugar fraction was analyzed by HPLC (Shodex SUGAR KS-801 column  $(300 \times 8.0$  mm, Showa Denko); H<sub>2</sub>O, 1.0 ml/min, chiral detection):  $t<sub>R</sub>$  7.5 min (p-glucose, positive optical rotation).

## **REFERENCES**

- [1] 'The Japanese Pharmacopoeia Fifteenth Edition', Ed. Society of Japanese Pharmacopoeia, Jiho, Tokyo, 2006, p. 1233.
- [2] H. Inouve, Y. Nakamura, *Tetrahedron Lett*. **1966**, 9, 4919.
- [3] H. Inouye, S. Ueda, Y. Nakamura, Tetrahedron Lett. 1966, 7, 5229.
- [4] Y. Ikeshiro, Y. Tomita, Planta Med. 1984, 50, 485.
- [5] Y. Ikeshiro, Y. Tomita, Planta Med. 1985, 51, 390.
- [6] Y. Ikeshiro, Y. Tomita, Planta Med. 1987, 53, 158.
- [7] T. Tomimori, M. Komatsu, Yakugaku Zasshi 1969, 89, 410.
- [8] I. Sakamoto, T. Tanaka, O. Tanaka, T. Tomimori, Chem. Pharm. Bull. 1982, 30, 4088.
- [9] P. Basnet, S. Kadota, M. Shimizu, T. Namba, Planta Med. 1994, 60, 507.
- [10] K. Hase, S. Kadota, P. Basnet, J. Li, S. Takamura, T. Namba, Chem. Pharm. Bull. 1997, 45, 567.
- [11] M. Komatsu, T. Tomimori, Y. Makiguchi, K. Asano, Yakugaku Zasshi 1968, 88, 832.
- [12] Y. Ikeshiro, T. Kubota, Y. Tomita, Planta Med. 1983, 47, 26.
- [13] Y. Tomita, J. Chem. Soc. Jpn. 1961, 82, 505.
- [14] T. Sakai, Y. Nakagawa, T. Iwashita, H. Naoki, T. Sakan, Bull. Chem. Soc. Jpn. 1983, 56, 3477.
- [15] M. Kikuchi, M. Kikuchi, Chem. Pharm. Bull. 2004, 52, 1210.
- [16] M. Kikuchi, M. Kikuchi, Chem. Pharm. Bull. 2005, 53, 48.
- [17] A. F. Beecham, *Tetrahedron* **1972**, 28, 5543.
- [18] R. D. Burnett, D. N. Kirk, J. Chem. Soc., Perkin Trans. 1 1981, 1460.
- [19] R. X. Tan, L. D. Kong, H. X. Wei, Phytochemistry 1998, 47, 1223.
- [20] Y. Liang, J. Hu, H. Chen, T. Zhang, Y. Ito, J. Liq. Chrom. Rel. Technol. 2007, 30, 509.
- [21] S.-S. Wang, W.-J. Zhao, X.-W. Han, X.-M. Liang, Chem. Pharm. Bull. 2005, 53, 674.
- [22] M. Kitajima, N. Fujii, F. Yoshino, H. Sudo, K. Saito, N. Aimi, H. Takayama, Chem. Pharm. Bull. 2005, 53, 1355.

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